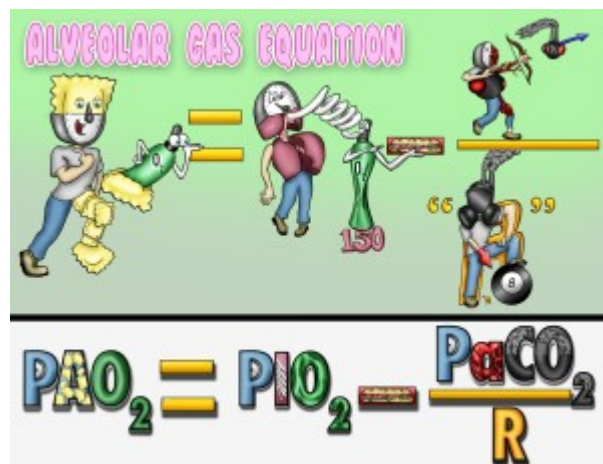


## Alveolar Gas Equation

The alveolar gas equation is used to calculate the alveolar partial pressure of oxygen (PAO<sub>2</sub>). This is done by the partial pressure of oxygen in the inspired air subtracted by the partial pressure of carbon dioxide in the arteries first divided by the respiratory quotient. The partial pressure of carbon dioxide divided by the respiratory quotient yields an estimate of the proportion of oxygen consumed by the body. Subtracting the partial pressure of oxygen in the air by the partial pressure of oxygen consumed by the body gives you the partial pressure of oxygen in the alveoli. The equation can be estimated as  $PAO_2 = 150 - (PaCO_2/0.8)$ . It is important to calculate the PAO<sub>2</sub> because this value is necessary in the calculation of alveolar-arterial gradient (A-a gradient), which is used clinically to diagnose the source of hypoxemia. The A-a gradient can be calculated by subtracting the alveolar partial pressure of oxygen from the arterial partial pressure of oxygen, which is normally 10 to 15 mmHg. Increased A-a gradient suggests possible shunting, VQ mismatch or fibrosis.



PLAY PICMONIC

### Equation

#### Partial Pressure of Alveolar Oxygen (PAO<sub>2</sub>)

[Partial Pressure-gauge with Ravioli O<sub>2</sub>-tank](#)

This is the partial pressure of oxygen (PO<sub>2</sub>) in the pulmonary alveoli, typically measured in millimeter of mercury (mmHg). This value is important for the calculation of other clinically relevant values, such as the A-a gradient.

#### Partial Pressure of Oxygen in the Inspired Air (PIO<sub>2</sub>)

[Partial Pressure-gauge Spiral-straw O<sub>2</sub>](#)

Oxygen exerts a partial pressure, which is determined by the environmental pressure. At sea level, the atmospheric pressure is 760 mmHg, and oxygen makes up 21 percent (20.94 percent to be exact) of inspired air. Since alveolar gas is assumed to be saturated with water vapor, the vapor pressure of water (~ 47 mmHg) must be subtracted from atmospheric pressure (760 - 47 = 713). So oxygen exerts a partial pressure of 713 x 0.21 = 150 mmHg.

#### PIO<sub>2</sub> Normally Approximated = 150 mmHg

[150 O<sub>2</sub>-tank](#)

PIO<sub>2</sub> in normal atmospheric settings is approximated as 150 mmHg.

#### Arterial Partial Pressure of CO<sub>2</sub> (PaCO<sub>2</sub>)

[Partial Pressure-gauge Artery-archer with CO<sub>2</sub>](#)

The arterial carbon dioxide partial pressure (PaCO<sub>2</sub>) is an indicator of carbon dioxide in the blood. A high PaCO<sub>2</sub>, or hypercapnia, is indicative of underventilation, or more rarely, hypermetabolic activity. A low PaCO<sub>2</sub>, or hypocapnia, is indicative of hyperventilation. This value can be obtained clinically via an arterial blood gas measurement (ABG).

#### Respiratory Quotient (R)

[Respirator Quotes](#)

The respiratory quotient is the amount of CO<sub>2</sub> produced, divided by the amount of O<sub>2</sub> consumed. This varies from organism to organism, but can be approximated at 0.8 for humans.

**R Normally Approximated = 0.8**

[Point \(8\) Ball](#)

Respiratory quotient is approximately 0.8 in humans.

**PAO<sub>2</sub> = PIO<sub>2</sub> - (PaCO<sub>2</sub>/R)**

[Ravioli with O<sub>2</sub>-tank EQUALS Spiral-straw with O<sub>2</sub>-tank SUB-tract Artery-Archer CO<sub>2</sub> OVER Respirator Quotes](#)

This is the equation used to calculate the alveolar partial pressure of oxygen. This equation can normally be approximated as PAO<sub>2</sub> = 150 - (PaCO<sub>2</sub>/0.8).